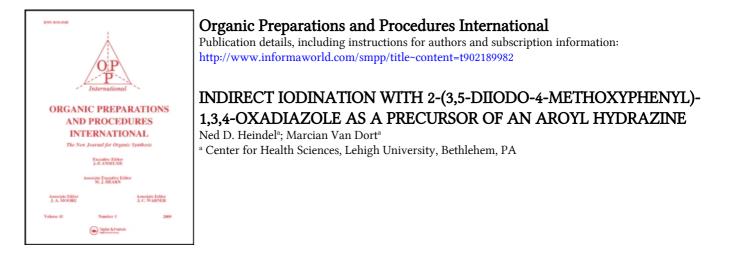
This article was downloaded by: On: *27 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



To cite this Article Heindel, Ned D. and Van Dort, Marcian(1985) 'INDIRECT IODINATION WITH 2-(3,5-DIIODO-4-METHOXYPHENYL)-1,3,4-OXADIAZOLE AS A PRECURSOR OF AN AROYL HYDRAZINE', Organic Preparations and Procedures International, 17: 3, 230 – 235

To link to this Article: DOI: 10.1080/00304948509355510 URL: http://dx.doi.org/10.1080/00304948509355510

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

chloride (40 ml). A saturated sodium chloride solution (30 ml) was added to the CH_2Cl_2 extracts. The sodium chloride solution was then reextracted twice with methylene chloride (40 ml). The combined organic extracts were dried (MgSO₄) and concentrated to give the crude sulfone. Analysis by TLC and NMR confirmed that the sulfide had been consumed. The known sulfones were purified by vacuum distillation or recrystallization and characterized by NMR and IR. The percent yields and physical properties are reported in Table 1.

REFERENCES

- 1. A. Pomerantz and R. Conner, J. Am. Chem. Soc., <u>61</u>, 3388 (1939).
- C. G. Venier, T. G. Squires, Y-Y. Chen., G. P. Hussmann, J. C. Shei and B. F. Smith, J. Org. Chem., 47, 3773 (1982).
- 3. R. L. Shriner, H. C. Struck and W. J. Jorison, J. Am. Chem. Soc., <u>52</u>, 2060 (1930).
- 4. D. Scholz, Montash. Chem., 112, 241 (1983).
- 5. R. U. Lemieux and E. von Rudolff, Can. J. Chem., 33, 1710 (1955).
- 6. H. B. Henbest and S. A. Khan, Chem. Commun., 1036 (1968).
- 7. E. von Rudolff, Can. J. Chem., 43, 1784 (1965).

INDIRECT IODINATION WITH 2-(3,5-DIIODO-4-METHOXYPHENYL)-

1,3,4-OXADIAZOLE AS A PRECURSOR OF AN AROYL HYDRAZINE

<u>Submitted by</u> Ned D. Heindel^{*} and Marcian Van Dort (01/28/85) Center for Health Sciences, Lehigh University Bethlehem, PA 18015

Iodinated contrast enhancement agents are frequently used to add

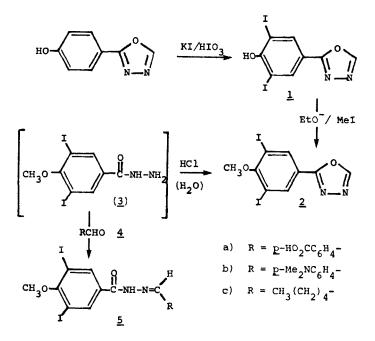
2011

Downloaded At: 11:24 27 January

OPPI BRIEFS

opacity to proximal tissues under radiographic scrutiny by computed tomography.¹ Hydroxyl groups on carrier molecules such as cholesterol or androgenic steroids have been used to attach iodinated benzoic acids and the resultant esters employed to enhance the contrast in tissues bearing steroidal receptors.² No convenient iodinated prosthetic group has been found for attachment at the carbonyl function of site-directed biomolecules.

We now report a facile route to an iodinated benzoic acid hydrazide $(\underline{3})$ suitable for this purpose. Earlier work demonstrated that halogens <u>ortho</u> to phenolic hydroxyls rapidly deiodinate <u>in vivo³</u> but this does not appear to be the case with <u>ortho</u>-halogenated anisoles.⁴



Direct iodination of N-H containing molecules has, however, often been thwarted by apparent oxidative side-reactions.^{5,6} These problems have been solved by the use of 3,5-diiodo-4-methoxybenzhydrazide ($\underline{3}$) which, however, was unobtainable by direct electrophilic iodinations (IC1 or KI, HIO₃) of either 4-hydroxy1- or 4-methoxybenzhydrazide. The N-H

functionalities of the hydrazide were therefore protected as a 1,3,4- oxadiazole, but even in this case iodination of 2-(4-methoxypheny1)-1,3,4-oxadiazole gave only modest yields of 2. On the other hand, 2-(4-hydroxypheny1)-1,3,4-oxadiazole could be iodinated in 66% yield and methylated in 77% yield to give 2. This oxadiazole serves as a shelf-stable, one-step precursor of 3,5-diiodo-4-methoxybenzhydrazide (3) which was generated and trapped <u>in situ</u> with carbonyl compounds to produce the hydrazones (5a-c) in 61-74% yields.

EXPERIMENTAL SECTION

NMR spectra were recorded on a JEOL-FX90Q instrument with TMS internal standard. IR spectra were determined on a Perkin-Elmer 283 spectrometer. Elemental analyses were performed by the Robertson Microanalytical Laboratory, Florham Park, NJ.

<u>2-(3,5-Diiodo-4-hydroxyphenyl)-1,3,4-oxadiazole</u> (1).- A solution of 10.0 g (62.0 mmol) of 2-(4-hydroxyphenyl)-1,3,4-oxadiazole,⁷ 10.3 g (62.0 mmol) of potassium iodide, 93 ml of 2N acetic acid (186 mmol), 40 ml of water and 76 ml of ethanol was refluxed with vigorous magnetic stirring for 15 min. Iodic acid (6.5 g, 37.0 mmol) in 50 ml of water was added over 0.5 hr, the mixture was stirred at reflux for 1 hr, and additional potassium iodide (10.3 g, 62.0 mmol in 40 ml of water) was added over a period of 15 min. After 1 hr of reflux, the solution was evaporated in vacuo to ca. one-fourth of its volume, chilled in an ice-water bath and the precipitated product collected and recrystallized from dioxane:water (1:2) to yield 17.0 g (66%) of 1, mp. 198-200°; IR: 3140 cm⁻¹ (OH); nmr (DMSO-d₆): δ 8.28 (s, 2, ArH), 9.29 (s, 1, oxadiazole C₅-H), and 10.28 ppm (br s, 1, OH).

<u>Anal</u>. Calcd. for C₈H₄I₂N₂O₂: C, 23.21; H, 0.97; N, 6.76 Found: C, 23.40; H, 1.08; N, 6.57

2-(3,5-Diiodo-4-methoxypheny1)-1,3,4-oxadiazole (2).- To a well-stirred suspension of 9.9 g (24.0 mmol), of 1, 1.84 g (27.0 mmol) of sodium

Downloaded At: 11:24 27 January 2011

OPPI BRIEFS

ethoxide, and 80 ml of ethanol:tetrahydrofuran (1:1) at reflux was added dropwise 5.68 g (40 mmol) of methyl iodide over a period of 15 min. The mixture was refluxed for 1 hr, 4.54 g (32 mmol) of methyl iodide added, and reflux continued for 2 hrs. After concentration to one-half volume <u>in vacuo</u>, the mixture was poured over chopped ice and the solid collected. Analytically pure material (7.9 g, 77% yield) was obtained by recrystallization from tetrahydrofuran-water (1:2), mp. $164-166^{\circ}$; nmr (DMSO-d₆): δ 3.81 (s, 3, OCH₃) 8.36 (s, 2, ArH), and 9.36 ppm (s, 1, oxadiazole C₅-H).

Anal. Calcd. for C₉H₆I₂N₂O₂: C, 25.25; H, 1.41; N, 6.55

Found: C, 25.14; H, 1.31; N, 6.28

<u>Preparation of Hydrazones</u>. <u>General Procedure</u>.- Hydrazones could be prepared in one-step by <u>in situ</u> hydrolysis of <u>2</u> and trapping of carbonyl components in the same reaction vessel or by isolation of the intermediate <u>3</u> and subsequent condensation with the aldehydes in a second step. Differences in overall yield were negligible. A solution of 6.0 g (14 mmol) of 2-(3,5-diiodo-4-methoxypheny1)-1,3,4-oxadiazole(<u>2</u>), 4.0 ml of conc. hydrochloric acid and 125 ml of tetrahydrofuran wasrefluxed for 3.5 hrs and evaporated to dryness in vacuo.

If at this point the residue was dissolved in 80 ml of water:THF (1:1), neutralized with conc. aqueous ammonia, reevaporated to dryness and recrystallized from water:THF (1:1), an 82% yield of analytically pure 3 could be obtained, mp. 191-192°; IR 3305/3200 (NH), 1630 cm⁻¹ (C=0); nmr (DMSO-d₆): δ 3.74 (s, 3, OCH₃), 4.49 (br s, 2, NH₂), 8.21 (s, 2, Ar-H), and 9.81 (br s, 1, CONH).

<u>Anal</u>. Calcd. for $C_8 H_8 I_2 N_2 O_2$: C, 22.99; H, 1.93; N, 6.70

Found: C, 23.26; H, 2.08; N, 6.63

This isolation procedure, however, was superfluous since the crude

233

hydrazide itself could be used directly in the syntheses of hydrazones $(\underline{5a-c})$ by addition of 14 mmol of aldehyde $(\underline{4a-c})$, 150 ml of ethanol:THF (1:1) and 1 ml of glacial acetic acid to the dry unpurified hydrazide residue. After 2 hrs of reflux with vigorous stirring, the mixture was evaporated to dryness and the hydrazones recrystallized to analytical purity as described below.

<u>3,5-Diiodo-4-methoxybenzoyl-(4-carboxyphenyl)hydrazone</u> (<u>5a</u>) obtained in 74% yield, mp. 272-273^o (from THF), by the direct combination of <u>2</u> and <u>4a</u>.

<u>Anal</u>. Calcd. for $C_{16}H_{12}I_2N_2O$: C, 34.93; H, 2.20; N, 5.09 Found: C, 34.69; H, 2.42; N, 5.07

<u>3,5-Diiodo-4-methoxybenzoyl-(4-N,N-dimethylaminophenyl)hydrazone</u> (5b) obtained in 62% yield, mp. $182-183^{\circ}$ from THF-hexane (1:3) by the direct combination of 2 and 4b.

<u>Anal</u>. Calcd. for C₁₇H₁₇I₂N₃O₂: C, 37.18, H, 3.12; N, 7.65

Found: C, 36.94; H, 3.16; N, 7.66

<u>3,5-Diiodo-4-methoxybenzoy1-(n-penty1)hydrazone</u> (<u>5c</u>) obtained in 61% yield, mp. 183-185⁰ [from THF:50-110⁰ pet. ether (1:3)] by direct combination of <u>2</u> and <u>4c</u> as described above.

<u>Anal</u>. Calcd. for C₁₃H₁₆I₂N₂O₂: C, 32.12; H, 3.32; N, 5.76

Found: C, 31.98; H, 3.24; N, 5.57

Acknowledgment. - This work was supported in part by grants from the Milheim Foundation, W. W. Smith Charitable Trust and DHHS (CA 31245).

REFERENCES

1. J. D. Meindl, Science, 215, 792 (1982).

- R. H. Seevers, M. P. Groziak, J. P. Weichert, S. W. Schwender, S. M. Szabo, M. A. Longino and R. E. Counsell, J. Med. Chem., <u>25</u>, 1500 (1982).
- 3. R. E. Counsell and R. D. Ice, "The Design of Organ Imaging

234

Radiopharmaceuticals", in "Drug Design", Vol. 6, E. J. Ariens, Ed., Academic Press, N.Y., N.Y. 1974, pp. 187, 197 and 209.

- Sargent, D. A. Kalbhen, A. T. Shulgin, H. Stauffer and N. Kusubou, J. Nucl. Med., 16, 243 (1975).
- 5. U. Braun, A. T. Shulgin, G. Braun and T. Sargent III, J. Med. Chem., 20, 1543 (1977).
- D. Futerte, E. Morier, M. Le Poncin-Lafitte, J. R. Rapin and R. Rips, J. Labelled Compds. and Radiopharm., 20, 149 (1983).
- 7. J. Maillard and M. Vincent. Bull. Soc. Chim. Fr., 376 (1966).

A CONVENIENT SYNTHESIS OF N-5-(1,1-DIMETHYLETHYL)-

3-ISOXAZOLYL-N, N-DIMETHYLUREA

Submitted by E. V. P. Tao* and G. S. Staten (11/05/84) The Lilly Research Laboratories Eli Lilly and Company Indianapolis, IN 4 6285

A practical and economical synthesis of N-5-(1,1-dimethylethyl)-3isoxazolyl-N,N-dimethylurea (4),¹ directly from pivalic acid without isolation of intermediates has been developed. This preparation offers the advantage that large scale reactions can be performed economically and efficiently

 $\begin{array}{c} \text{Me}_{3}\text{CCO}_{2}\text{H} \xrightarrow{i:\text{BuOH}} & \text{Me}_{3}\text{CCO}_{2}\text{-}i:\text{Bu} \xrightarrow{\text{CH}_{3}\text{CN}} & \text{Me}_{3}\text{CCOCH}_{2}\text{CN} \xrightarrow{\text{H}_{2}\text{NOH}} \\ & \begin{array}{c} 1 \\ 1 \\ \end{array} & \begin{array}{c} 2 \\ \end{array} \\ & \begin{array}{c} 1 \\ \end{array} & \begin{array}{c} 2 \\ \end{array} \\ & \begin{array}{c} \text{NHCONMe}_{2} \\ \end{array} \\ & \begin{array}{c} \text{Me}_{3}\text{C} \xrightarrow{\text{O}} \\ \end{array} \\ & \begin{array}{c} 3 \\ \end{array} \end{array} \end{array} \\ \begin{array}{c} \text{Me}_{3}\text{C} \xrightarrow{\text{O}} \\ \end{array} \\ & \begin{array}{c} 4 \\ \end{array} \end{array}$